

# SAFETY DATA SHEETS

According to the UN GHS revision 9

Version: 1.0  
Creation Date: July 15, 2019  
Revision Date: July 15, 2019

## SECTION 1: Identification

### 1.1 GHS Product identifier

**Product name** Calcium hypochlorite

### 1.2 Other means of identification

**Product number**

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**Other names**

Hypochlorous acid calcium salt;

### 1.3 Recommended use of the chemical and restrictions on use

**Identified uses**

Industrial and scientific research use.

**Uses advised against**

no data available

### 1.4 Supplier's details

**Company**

Shanghai Baishun Biotechnology Co., Ltd

**Address**

No. 26, Lane 918, Lianye Road, Zhelin Town, Fengxian District, Shanghai, 201400, China

**Telephone**

+86-21-37581181

### 1.5 Emergency phone number

**Emergency phone number**

+86-21-37581181

**Service hours**

Monday to Friday, 9am-5pm (Standard time zone: UTC/GMT +8 hours).

## SECTION 2: Hazard identification

### 2.1 Classification of the substance or mixture

Oxidizing solids, Category 2

Acute toxicity - Category 4, Oral

Skin corrosion, Sub-category 1B

Hazardous to the aquatic environment, short-term (Acute) - Category Acute 1

### 2.2 GHS label elements, including precautionary statements

**Pictogram(s)**





<b>Signal word</b>	Danger
<b>Hazard statement(s)</b>	H272 May intensify fire; oxidizer H302 Harmful if swallowed H314 Causes severe skin burns and eye damage H400 Very toxic to aquatic life
<b>Precautionary statement(s)</b>	
<b>Prevention</b>	P210 Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking. P220 Keep away from clothing and other combustible materials. P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/... P264 Wash ... thoroughly after handling. P270 Do not eat, drink or smoke when using this product. P260 Do not breathe dust/fume/gas/mist/vapours/spray. P273 Avoid release to the environment.
<b>Response</b>	P370+P378 In case of fire: Use ... to extinguish. P301+P317 IF SWALLOWED: Get medical help. P330 Rinse mouth. P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting. P363 Wash contaminated clothing before reuse. P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing. P316 Get emergency medical help immediately. P321 Specific treatment (see ... on this label). P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. P391 Collect spillage.
<b>Storage</b>	P405 Store locked up.
<b>Disposal</b>	P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

## 2.3 Other hazards which do not result in classification

no data available

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## SECTION 3: Composition/information on ingredients

### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Calcium hypochlorite	Calcium hypochlorite	7778-54-3	231-908-7	100%

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## SECTION 4: First-aid measures

### 4.1 Description of necessary first-aid measures

#### If inhaled

Fresh air, rest. Half-upright position. Artificial respiration may be needed. Refer for medical attention.

#### Following skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower.

#### Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

#### **Following ingestion**

Rinse mouth. Give one or two glasses of water to drink. Do NOT induce vomiting. Refer for medical attention .

### **4.2 Most important symptoms/effects, acute and delayed**

Excerpt from ERG Guide 140 [Oxidizers]: Inhalation, ingestion or contact (skin, eyes) with vapors or substance may cause severe injury, burns or death. Fire may produce irritating, corrosive and/or toxic gases. Runoff from fire control or dilution water may cause pollution. (ERG, 2016)

### **4.3 Indication of immediate medical attention and special treatment needed, if necessary**

Immediate first aid: Ensure that adequate decontamination has been carried out. If patient is not breathing, start artificial respiration, preferably with a demand-valve resuscitator, bag-valve-mask device, or pocket mask, as trained. Perform CPR as necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep patient quiet and maintain normal body temperature. Obtain medical attention. Hypochlorite and Related Compounds

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## **SECTION 5: Fire-fighting measures**

### **5.1 Suitable extinguishing media**

If material involved in fire: Extinguish using agent suitable for type of surrounding fire. (Material itself does not burn or burns with difficulty.) Use water in flooding quantities as fog. Cool all affected containers with flooding quantities of water. Apply water from as far a distance as possible. Chlorinated lime, liquid

### **5.2 Specific hazards arising from the chemical**

Excerpt from ERG Guide 140 [Oxidizers]: These substances will accelerate burning when involved in a fire. Some may decompose explosively when heated or involved in a fire. May explode from heat or contamination. Some will react explosively with hydrocarbons (fuels). May ignite combustibles (wood, paper, oil, clothing, etc.). Containers may explode when heated. Runoff may create fire or explosion hazard. (ERG, 2016)

### **5.3 Special protective actions for fire-fighters**

NO powder. Use water in large amounts. In case of fire: keep drums, etc., cool by spraying with water.

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## **SECTION 6: Accidental release measures**

### **6.1 Personal precautions, protective equipment and emergency procedures**

Personal protection: chemical protection suit, face shield and filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Sweep spilled substance into covered air-tight, dry containers. Then store and dispose of according to local regulations.

### **6.2 Environmental precautions**

Personal protection: chemical protection suit, face shield and filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Sweep spilled substance into covered air-tight, dry containers. Then store and dispose of according to local regulations.

### **6.3 Methods and materials for containment and cleaning up**

Personal protection: chemical protection suit, face shield and filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Do NOT let this

chemical enter the environment. Sweep spilled substance into covered air-tight, dry containers. Then store and dispose of according to local regulations.

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## SECTION 7: Handling and storage

### 7.1 Precautions for safe handling

NO contact with combustible substances or reducing agents. Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

### 7.2 Conditions for safe storage, including any incompatibilities

Well closed. Store in an area without drain or sewer access. Separated from food and feedstuffs. See Chemical Dangers. Store in a cool, dry, well-ventilated location at a temperature below 120 deg F (50 deg C) to avoid slow decomposition. Separate from oxidizing materials, acids, ammonia, amines, and other chlorinating agents. Immediately remove and properly dispose of any spilled material. Calcium hypochlorite, dry or calcium hypochlorite, mixtures, dry

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## SECTION 8: Exposure controls/personal protection

### 8.1 Control parameters

#### Occupational Exposure limit values

no data available

#### Biological limit values

no data available

### 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

### 8.3 Individual protection measures, such as personal protective equipment (PPE)

#### Eye/face protection

Wear face shield or eye protection in combination with breathing protection.

#### Skin protection

Protective gloves. Protective clothing.

#### Respiratory protection

Use local exhaust or breathing protection.

#### Thermal hazards

no data available

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## SECTION 9: Physical and chemical properties and safety characteristics

Physical state	Solid. Crystalline.
Colour	White.
Odour	Strong chlorine odor
Melting point/freezing point	100 (decomposition)
Boiling point or initial boiling point and boiling range	Decomposes at 100 deg C
Flammability	Not combustible but enhances combustion of other substances. Many reactions may cause fire or explosion. Gives off irritating or toxic fumes (or gases) in a fire.

<b>Lower and upper explosion limit/flammability limit</b>	no data available
<b>Flash point</b>	no data available
<b>Auto-ignition temperature</b>	no data available
<b>Decomposition temperature</b>	100°C
<b>pH</b>	no data available
<b>Kinematic viscosity</b>	no data available
<b>Solubility</b>	Miscible with water
<b>Partition coefficient n-octanol/water</b>	no data available
<b>Vapour pressure</b>	no data available
<b>Density and/or relative density</b>	2.35
<b>Relative vapour density</b>	no data available
<b>Particle characteristics</b>	no data available

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## SECTION 10: Stability and reactivity

### 10.1 Reactivity

Decomposes rapidly above 175°C . Decomposes rapidly on contact with acids. This produces chlorine and oxygen. This generates fire and explosion hazard. The substance is a strong oxidant. It reacts violently with combustible and reducing materials. The solution in water is a medium strong base. Reacts violently with ammonia, amines, nitrogen compounds and many other substances. This generates explosion hazard. Attacks many metals. This produces flammable/explosive gas (hydrogen - see ICSC 0001). Attacks plastics.

### 10.2 Chemical stability

All hypochlorite soln are unstable, esp if acidified; slowly decomp on contact with air hypochlorite soln

### 10.3 Possibility of hazardous reactions

Contact with combustible materials will increase fire hazard. May undergo accelerated decomposition with release of heat above 350 deg F (177 deg C). /Calcium hypochlorite, dry, or calcium hypochlorite, mixtures, dry/CALCIUM HYPOCHLORITE is a powerful oxidizing agent, particularly in the presence of water or as it decomposes when heated to release oxygen and chlorine gases. May react vigorously with carbon; reacts potentially explosively with finely divided carbon. Reacts with acetylene to form explosive chloroacetylenes. Reactions with organic matter, oil, hydrocarbons; alcohols may lead to explosions. Reactions with nitromethane, methanol, ethanol (and other alcohols) can become violent after a delay. Reacts with possible ignition and/or explosion with organic sulfur compounds and with sulfides. Decomposes evolving oxygen, a change that can be catalyzed by rust on metal containers. Forms highly explosive NCl<sub>3</sub> on contact with urea or ammonia. Evolves highly toxic gaseous chlorine gas when heated or on contact with acids [Sax, 9th ed., 1996, p. 1905]. A mixture with damp sulfur reacted violently, and molten sulfur was ejected [Chem Eng. News, 1965, 46(29), 6]. The combination of calcium hypochlorite, sodium hydrogen sulfate, starch, and sodium carbonate, when compressed, caused the materials to incandescence, followed by explosion, [Ind. Eng. Chem., 1937, 15, 282].

### 10.4 Conditions to avoid

no data available

### 10.5 Incompatible materials

Reacts with water and with acids releasing chlorine. Forms explosive compounds with ammonia and amines. Strong oxidizer. Other incompatible materials include organics, nitrogen containing compounds, dry chemical fire extinguishers containing mono-ammonium phosphate, combustible or flammable materials. Calcium hypochlorite, dry or calcium hypochlorite, mixtures, dry

## 10.6 Hazardous decomposition products

The 70% grade may decomp violently if exposed to heat or direct sunlight. Gives off chlorine and chlorine monoxide above 350 deg F (poisonous gases).

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## SECTION 11: Toxicological information

### Acute toxicity

- Oral: LD50 Rat (male) oral 790 mg/kg
- Inhalation: no data available
- Dermal: no data available

### Skin corrosion/irritation

no data available

### Serious eye damage/irritation

no data available

### Respiratory or skin sensitization

no data available

### Germ cell mutagenicity

no data available

### Carcinogenicity

Evaluation: There is inadequate evidence for the carcinogenicity of hypochlorite salts in experimental animals. No data were available from studies in humans on the carcinogenicity of hypochlorite salts. Overall evaluation: Hypochlorite salts are not classifiable as to their carcinogenicity to humans (Group 3). Hypochlorite salts

### Reproductive toxicity

no data available

### STOT-single exposure

The substance is corrosive to the eyes, skin and respiratory tract. Corrosive on ingestion. Inhalation of decomposition products may cause lung oedema. See Notes. The effects may be delayed. Medical observation is indicated.

### STOT-repeated exposure

no data available

### Aspiration hazard

A harmful concentration of airborne particles can be reached quickly when dispersed, especially if powdered.

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## SECTION 12: Ecological information

### 12.1 Toxicity

- Toxicity to fish: LC50; Species: *Lepomis macrochirus* (Bluegill) juvenile, weight 3.91 g, length 4.6 cm; Conditions: freshwater, renewal, 32 deg C, pH 7.25-7.55, hardness 41.5-46.3 mg/L CaCO<sub>3</sub>, alkalinity 42-45 mg/L CaCO<sub>3</sub>; Concentration: 71 ug/L for 24 hr (95% confidence interval: 68-75 ug/L) /total Cl ion
- Toxicity to daphnia and other aquatic invertebrates: EC50; Species: *Daphnia magna* (Water flea) 1st instar; Conditions: freshwater, static; Concentration: 73 ug/L for 48 hr (95% confidence interval: 67-79 ug/L); Effect: intoxication, immobilization /65% purity
- Toxicity to algae: EC50; Species: *Pseudokirchneriella subcapitata* (Green algae) 3 day algal culture, 1,200,000 cells/cu cm; Conditions: freshwater, 27 deg C; Concentration: 983 ug/L for 72 hr (95% confidence interval: 805-1210 ug/L); Effect: decreased population growth rate /Chlorinated lime
- Toxicity to microorganisms: no data available

## 12.2 Persistence and degradability

no data available

## 12.3 Bioaccumulative potential

no data available

## 12.4 Mobility in soil

no data available

## 12.5 Other adverse effects

no data available

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# SECTION 13: Disposal considerations

## 13.1 Disposal methods

### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

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# SECTION 14: Transport information

## 14.1 UN Number

ADR/RID: UN1748 (For reference only, please check.)

IMDG: UN1748 (For reference only, please check.)

IATA: UN1748 (For reference only, please check.)

## 14.2 UN Proper Shipping Name

ADR/RID: CALCIUM  
HYPOCHLORITE, DRY or  
CALCIUM  
HYPOCHLORITE  
MIXTURE, DRY with more  
than 39% available chlorine  
(8.8% available oxygen) (For  
reference only, please check.)

IMDG: CALCIUM  
HYPOCHLORITE, DRY or  
CALCIUM  
HYPOCHLORITE  
MIXTURE, DRY with more  
than 39% available chlorine  
(8.8% available oxygen) (For  
reference only, please check.)

IATA: CALCIUM  
HYPOCHLORITE, DRY or  
CALCIUM  
HYPOCHLORITE  
MIXTURE, DRY with more  
than 39% available chlorine  
(8.8% available oxygen) (For  
reference only, please check.)

## 14.3 Transport hazard class(es)

ADR/RID: 5.1 (For reference only, please check.)

IMDG: 5.1 (For reference only, please check.)

IATA: 5.1 (For reference only, please check.)

## 14.4 Packing group, if applicable

ADR/RID: II (For reference only, please check.)

IMDG: II (For reference only, please check.)

IATA: II (For reference only, please check.)

## 14.5 Environmental hazards

ADR/RID: Yes

IMDG: Yes

IATA: Yes

## 14.6 Special precautions for user

no data available

## 14.7 Transport in bulk according to IMO instruments

no data available

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## SECTION 15: Regulatory information

### 15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
Calcium hypochlorite	Calcium hypochlorite	7778-54-3	231-908-7
European Inventory of Existing Commercial Chemical Substances (EINECS)			Listed.
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Listed.
New Zealand Inventory of Chemicals (NZIoC)			Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			Listed.
Vietnam National Chemical Inventory			Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			Listed.
Korea Existing Chemicals List (KECL)			Listed.

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## SECTION 16: Other information

### Information on revision

**Creation Date** July 15, 2019

**Revision Date** July 15, 2019

### Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

### References

- IPCS - The International Chemical Safety Cards (ICSC), website: <http://www.ilo.org/dyn/icsc/showcard.home>
- HSDB - Hazardous Substances Data Bank, website: <https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm>
- IARC - International Agency for Research on Cancer, website: <http://www.iarc.fr/>
- eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: [http://www.echemportal.org/echemportal/index?pageID=0&request\\_locale=en](http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en)
- CAMEO Chemicals, website: <http://cameochemicals.noaa.gov/search/simple>
- ChemIDplus, website: <http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp>
- ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: <http://www.phmsa.dot.gov/hazmat/library/erg>
- Germany GESTIS-database on hazard substance, website: <http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp>
- ECHA - European Chemicals Agency, website: <https://echa.europa.eu/>

### Other Information

Rinse contaminated clothing with plenty of water because of fire hazard. The symptoms of lung oedema often do not become manifest until a few hours have passed and they are aggravated by physical effort. Rest and medical observation is therefore essential. Immediate



administration of an appropriate inhalation therapy by a doctor or a person authorized by him/her, should be considered.

**Any questions regarding this SDS, Please send your inquiry to [sds@xixisys.com](mailto:sds@xixisys.com)**

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