

# SAFETY DATA SHEETS

According to the UN GHS revision 9

Version: 1.0  
Creation Date: July 15, 2019  
Revision Date: July 15, 2019

## SECTION 1: Identification

### 1.1 GHS Product identifier

**Product name** Dibenzoyl peroxide

### 1.2 Other means of identification

**Product number** -  
**Other names** benzoyl benzenecarboperoxoate; Peroxide, dibenzoyl

### 1.3 Recommended use of the chemical and restrictions on use

**Identified uses** Industrial and scientific research use.  
**Uses advised against** no data available

### 1.4 Supplier's details

**Company** Shanghai Baishun Biotechnology Co., Ltd  
**Address** No. 26, Lane 918, Lianye Road, Zhelin Town, Fengxian District, Shanghai, 201400, China  
**Telephone** +86-21-37581181

### 1.5 Emergency phone number

**Emergency phone number** +86-21-37581181  
**Service hours** Monday to Friday, 9am-5pm (Standard time zone: UTC/GMT +8 hours).

## SECTION 2: Hazard identification

### 2.1 Classification of the substance or mixture

Organic peroxides, Type B  
Eye irritation, Category 2  
Skin sensitization, Category 1

### 2.2 GHS label elements, including precautionary statements

**Pictogram(s)**



**Signal word** Danger  
**Hazard statement(s)** H241 Heating may cause a fire or explosion  
H319 Causes serious eye irritation  
H317 May cause an allergic skin reaction

**Precautionary statement(s)**

<b>Prevention</b>	P210 Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking. P234 Keep only in original packaging. P235 Keep cool. P240 Ground and bond container and receiving equipment. P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/... P264 Wash ... thoroughly after handling. P261 Avoid breathing dust/fume/gas/mist/vapours/spray. P272 Contaminated work clothing should not be allowed out of the workplace.
<b>Response</b>	P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. P302+P352 IF ON SKIN: Wash with plenty of water/... P333+P317 If skin irritation or rash occurs: Get medical help. P321 Specific treatment (see ... on this label). P362+P364 Take off contaminated clothing and wash it before reuse.
<b>Storage</b>	P403 Store in a well-ventilated place. P410 Protect from sunlight. P411 Store at temperatures not exceeding ...°C/...°F. P420 Store separately.
<b>Disposal</b>	P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

### 2.3 Other hazards which do not result in classification

no data available

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## SECTION 3: Composition/information on ingredients

### 3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Dibenzoyl peroxide	Dibenzoyl peroxide	94-36-0	202-327-6	100%

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## SECTION 4: First-aid measures

### 4.1 Description of necessary first-aid measures

#### If inhaled

Fresh air, rest.

#### Following skin contact

Remove contaminated clothes. Rinse and then wash skin with water and soap.

#### Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

#### Following ingestion

Rinse mouth. Give one or two glasses of water to drink. Refer for medical attention .

### 4.2 Most important symptoms/effects, acute and delayed

CONTACT WITH EYES OR SKIN: irritates eyes. Prolonged contact may irritate skin. (USCG, 1999)  
Excerpt from ERG Guide 145 [Organic Peroxides (Heat and Contamination Sensitive)]:  
Fire may produce irritating, corrosive and/or toxic gases. Ingestion or contact (skin, eyes) with substance may cause severe injury or burns. Runoff from fire control or dilution water may cause pollution. (ERG, 2016)

Excerpt from ERG Guide 146 [Organic Peroxides (Heat, Contamination and Friction Sensitive)]: Fire may produce irritating, corrosive and/or toxic gases. Ingestion or contact (skin, eyes) with substance may cause severe injury or burns. Runoff from fire control or dilution water may cause pollution. (ERG, 2016)

#### **4.3 Indication of immediate medical attention and special treatment needed, if necessary**

For immediate first aid: Ensure that adequate decontamination has been carried out. If victim is not breathing, start artificial respiration, preferably with a demand-valve resuscitator, bag-valve-mask device, or pocket mask as trained. Perform CPR if necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep victim quiet and maintain normal body temperature. Obtain medical attention. Organic peroxides

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### **SECTION 5: Fire-fighting measures**

#### **5.1 Suitable extinguishing media**

If material on fire or involved in fire: Burns with explosive violence. If it becomes ignited, material cannot be extinguished until it is consumed. Benzoyl peroxide (benzoyl peroxide, dry or wet with less than 30% water)

#### **5.2 Specific hazards arising from the chemical**

Special Hazards of Combustion Products: Suffocating smoke evolved Behavior in Fire: May explode (USCG, 1999)

Excerpt from ERG Guide 145 [Organic Peroxides (Heat and Contamination Sensitive)]: May explode from heat or contamination. May ignite combustibles (wood, paper, oil, clothing, etc.). May be ignited by heat, sparks or flames. May burn rapidly with flare-burning effect. Containers may explode when heated. Runoff may create fire or explosion hazard. (ERG, 2016)

Excerpt from ERG Guide 146 [Organic Peroxides (Heat, Contamination and Friction Sensitive)]: May explode from heat, shock, friction or contamination. May ignite combustibles (wood, paper, oil, clothing, etc.). May be ignited by heat, sparks or flames. May burn rapidly with flare-burning effect. Containers may explode when heated. Runoff may create fire or explosion hazard. (ERG, 2016)

#### **5.3 Special protective actions for fire-fighters**

Use water in large amounts. In case of fire: keep drums, etc., cool by spraying with water. Combat fire from a sheltered position.

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### **SECTION 6: Accidental release measures**

#### **6.1 Personal precautions, protective equipment and emergency procedures**

Evacuate danger area! Remove all ignition sources. Wash away remainder with plenty of water. Do NOT absorb in saw-dust or other combustible absorbents.

#### **6.2 Environmental precautions**

Evacuate danger area! Remove all ignition sources. Wash away remainder with plenty of water. Do NOT absorb in saw-dust or other combustible absorbents.

#### **6.3 Methods and materials for containment and cleaning up**

Cover the spill with a 1:1:1 mixture by weight of sodium carbonate or calcium carbonate, clay cat litter (bentonite) and sand. Dampen this mixture thoroughly with water, then scoop into a beaker using a plastic or cardboard shovel. Treat as per waste disposal procedure. Wash the area of the spill thoroughly with soap and water.

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### **SECTION 7: Handling and storage**

#### **7.1 Precautions for safe handling**

NO open flames, NO sparks and NO smoking. NO contact with flammables. NO contact with hot surfaces. Do NOT expose to friction or shock. Use non-sparking handtools. Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with

skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

## 7.2 Conditions for safe storage, including any incompatibilities

Fireproof. Separated from combustible substances and reducing agents. Store only in original packaging. See Chemical Dangers. All precautions must be taken to guard against fire and explosion hazards. Keep in a cool place out of direct rays of the sun; away from sparks, open flames and other sources of heat; away from shock, rough handling, friction from grinding, etc. Isolated storage is required; keep away from possible contact with acids, alcohols, ethers or other reducing agents or polymerization catalysts such as dimethylaniline. Complete instructions on storage and handling available from manufacturer.

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## SECTION 8: Exposure controls/personal protection

### 8.1 Control parameters

#### Occupational Exposure limit values

TLV: 5 mg/m<sup>3</sup>, as TWA; A4 (not classifiable as a human carcinogen). MAK: 5 mg/m<sup>3</sup>; peak limitation category: I(1)

#### Biological limit values

no data available

### 8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

### 8.3 Individual protection measures, such as personal protective equipment (PPE)

#### Eye/face protection

Wear safety goggles.

#### Skin protection

Protective gloves. Protective clothing.

#### Respiratory protection

Use local exhaust or breathing protection.

#### Thermal hazards

no data available

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## SECTION 9: Physical and chemical properties and safety characteristics

Physical state	Solid. Crystals or granular powder.
Colour	White.
Odour	Faint, benzaldehyde-like odor.
Melting point/freezing point	105 °C. Atm. press.:1.013 hPa. Remarks:As no information are given for the atmospheric pressure, it is assumed to be at normal pressure.
Boiling point or initial boiling point and boiling range	176°F
Flammability	Combustible Solid (easily ignited and burns very rapidly).
Lower and upper explosion limit/flammability limit	no data available
Flash point	93 °C
Auto-ignition temperature	80 °C. Atm. press.:1.013 hPa.

<b>Decomposition temperature</b>	103-105°C
<b>pH</b>	no data available
<b>Kinematic viscosity</b>	no data available
<b>Solubility</b>	Insoluble in water
<b>Partition coefficient n-octanol/water</b>	log Pow = 3.2. Temperature:22 °C.
<b>Vapour pressure</b>	0.009 Pa. Temperature:25 °C. Remarks:Modified Grain Method.
<b>Density and/or relative density</b>	1.33. Temperature:25 °C.
<b>Relative vapour density</b>	no data available
<b>Particle characteristics</b>	no data available

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## SECTION 10: Stability and reactivity

### 10.1 Reactivity

May decompose explosively on shock, friction or concussion. May explode on heating above 103-105°C. On combustion, forms irritating and toxic fumes and gases of benzoic acid (see ICSC 0103) and carbon monoxide (see ICSC 0023). The substance is a strong oxidant. It reacts violently with combustible and reducing materials. Reacts violently with many organic acids, inorganic acids, alcohols and amines. This generates fire and explosion hazard.

### 10.2 Chemical stability

Very stable /srp: shelf life/ @ ordinary temp (25 deg c)

### 10.3 Possibility of hazardous reactions

Unstable to heat; decomposes with evolving oxygen, resulting in advancing fire. Sensitive to impact and friction. BENZOYL PEROXIDE is a white, odorless powder, moderately toxic. It is most dangerous when it contains less than 1% water. A moisture content of 3% allowed slow burning only, and at 5% ignition did not occur [McCloskey, C. M. et al., Chem. Abs., 1967, 66, 12613c]. Mixed with a large surplus of water (30% or more), it is relatively safe. In dry form, it is a very dangerous material. It will explode spontaneously when heated above melting point (103° C). An explosion which occurred when a screw-capped bottle of the peroxide was opened has been attributed to friction, which initiated a mixture of peroxide and organic dust in the cap threads [Lappin, G. R., Chem. Eng. News, 1948, 26, p.3518]. A violent explosion occurred during purification of the peroxide by Soxhlet extraction with hot chloroform [Anon., Sichere Chemiearb., 1976, 28, p. 49]. It is a powerful oxidizer, which ignites readily and burns rapidly. In contact with reducing agents, it may ignite by spontaneous chemical reaction. It must be kept in a cool place, in isolation and out of the sunlight or sources of heat. Also, avoid shock or friction. It reacts violently with inorganic or organic acids, alcohols, amines, metallic naphthenates and polymerization accelerators (e.g., N,N-dimethylaniline). Explosive or violent reaction upon contact with dimethyl sulfide, lithium aluminum hydride or aniline [Bretherick, 5th ed., 1995, p. 1140]. Mixture with carbon tetrachloride and ethylene explodes when exposed to heat [Bolt, R. O. et al., Chem. Eng. News, 1947, 25, p. 1866]. Ignition occurred upon contact with methyl methacrylate [MCA Case History No. 996], polymerization of vinyl acetate in ethyl acetate accelerated out of control leading to ignition and explosion [Vervalin, 1973, p. 81]. At 50° C a mixture of dibenzoyl peroxide and charcoal reacts violently producing dense white smoke of benzoic acid, benzene, phenyls and carbon dioxide [Leleu, Cahiers, 1980, 99, p. 279].

### 10.4 Conditions to avoid

no data available

### 10.5 Incompatible materials

Special care must be taken to avoid contamination with combustible materials (wood, paper), various inorganic & organic acids, alkalies, alcohols, amines, easily oxidizable materials such as ethers, or materials used as accelerators in polymerization reactions.

### 10.6 Hazardous decomposition products

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Explosive decomposition above the melting point ... forms flammable products.

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## SECTION 11: Toxicological information

### Acute toxicity

- Oral: LD0 - mouse (male/female) - > 2 000 mg/kg bw. Remarks: No mortality.
- Inhalation: no data available
- Dermal: no data available

### Skin corrosion/irritation

no data available

### Serious eye damage/irritation

no data available

### Respiratory or skin sensitization

no data available

### Germ cell mutagenicity

no data available

### Carcinogenicity

Evaluation: There is inadequate evidence in humans for the carcinogenicity of benzoyl peroxide. There is limited evidence in experimental animals for the carcinogenicity of benzoyl peroxide. Overall evaluation: Benzoyl peroxide is not classifiable as to its carcinogenicity to humans (Group 3).

### Reproductive toxicity

no data available

### STOT-single exposure

The substance is irritating to the eyes, skin and respiratory tract.

### STOT-repeated exposure

Repeated or prolonged contact may cause skin sensitization.

### Aspiration hazard

Evaporation at 20°C is negligible; a harmful concentration of airborne particles can, however, be reached quickly when dispersed, especially if powdered.

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## SECTION 12: Ecological information

### 12.1 Toxicity

- Toxicity to fish: LC50 - *Oncorhynchus mykiss* (previous name: *Salmo gairdneri*) - 0.06 mg/L - 96 h.
- Toxicity to daphnia and other aquatic invertebrates: EC50 - *Daphnia magna* - 0.11 mg/L - 48 h.
- Toxicity to algae: EC50 - *Pseudokirchneriella subcapitata* (previous names: *Raphidocelis subcapitata*, *Selenastrum capricornutum*) - 0.071 mg/L - 72 h.
- Toxicity to microorganisms: EC50 - activated sludge of a predominantly domestic sewage - 35 mg/L - 30 min.

### 12.2 Persistence and degradability

In the Japanese MITI test, benzoyl peroxide (present at 100 ppm) reached 84 percent of its theoretical BOD in 21 days, using an activated sludge inoculum(1).

### 12.3 Bioaccumulative potential

An estimated BCF of 250 was calculated for benzoyl peroxide(SRC), using an experimental log Kow of 3.46(1, SRC) and a recommended regression-derived equation(2). According to a classification scheme(3), this BCF suggests that bioconcentration in aquatic organisms is high(SRC).

## 12.4 Mobility in soil

The Koc of benzoyl peroxide is estimated as approximately 1,800(SRC), using a measured log Kow of 3.46(1) and a regression-derived equation(2,SRC). According to a recommended classification scheme(3), this estimated Koc value suggests that benzoyl peroxide is expected to have low mobility in soil(SRC).

## 12.5 Other adverse effects

no data available

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# SECTION 13: Disposal considerations

## 13.1 Disposal methods

### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

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# SECTION 14: Transport information

## 14.1 UN Number

ADR/RID: UN3108 (For reference only, please check.)

IMDG: UN3108 (For reference only, please check.)

IATA: UN3108 (For reference only, please check.)

## 14.2 UN Proper Shipping Name

ADR/RID: ORGANIC PEROXIDE TYPE E, SOLID (For reference only, please check.)

IMDG: ORGANIC PEROXIDE TYPE E, SOLID (For reference only, please check.)

IATA: ORGANIC PEROXIDE TYPE E, SOLID (For reference only, please check.)

## 14.3 Transport hazard class(es)

ADR/RID: 5.2 (For reference only, please check.)

IMDG: 5.2 (For reference only, please check.)

IATA: 5.2 (For reference only, please check.)

## 14.4 Packing group, if applicable

ADR/RID: (For reference only, please check.)

IMDG: (For reference only, please check.)

IATA: (For reference only, please check.)

## 14.5 Environmental hazards

ADR/RID: No

IMDG: No

IATA: No

## 14.6 Special precautions for user

no data available

## 14.7 Transport in bulk according to IMO instruments

no data available

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# SECTION 15: Regulatory information

## 15.1 Safety, health and environmental regulations specific for the product in question

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Chemical name	Common names and synonyms	CAS number	EC number
Dibenzoyl peroxide	Dibenzoyl peroxide	94-36-0	202-327-6
European Inventory of Existing Commercial Chemical Substances (EINECS)			Listed.
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Listed.
New Zealand Inventory of Chemicals (NZIoC)			Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			Listed.
Vietnam National Chemical Inventory			Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			Listed.
Korea Existing Chemicals List (KECL)			Listed.

## SECTION 16: Other information

### Information on revision

**Creation Date** July 15, 2019

**Revision Date** July 15, 2019

### Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

### References

- IPCS - The International Chemical Safety Cards (ICSC), website: <http://www.ilo.org/dyn/icsc/showcard.home>
- HSDB - Hazardous Substances Data Bank, website: <https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm>
- IARC - International Agency for Research on Cancer, website: <http://www.iarc.fr/>
- eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: [http://www.echemportal.org/echemportal/index?pageID=0&request\\_locale=en](http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en)
- CAMEO Chemicals, website: <http://cameochemicals.noaa.gov/search/simple>
- ChemIDplus, website: <http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp>
- ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: <http://www.phmsa.dot.gov/hazmat/library/erg>
- Germany GESTIS-database on hazard substance, website: <http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp>
- ECHA - European Chemicals Agency, website: <https://echa.europa.eu/>

### Other Information

Carrier solvents used in commercial formulations may change physical and toxicological properties. Rinse contaminated clothing with plenty of water because of fire hazard.

**Any questions regarding this SDS, Please send your inquiry to [sds@xixisys.com](mailto:sds@xixisys.com)**

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